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Kerala Plus Two Chemistry Chapter Wise Questions and Answers Chapter 11 Alcohols, Phenols and Ethers. Very Short Answer Type Questions (Score 1) Question 1. Glycerol is _____. Answer: 2, 3, 1 dihydroxy propanal. Question 2. Picric acid is a. 2, 4, 6 – trinitrotoluene

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Kerala Plus Two Chemistry Notes Chapter 3 Electrochemistry. Electrochemistry. It is the study of production of electricity from I energy which is released during spontaneous chemical reactions, also the use of electrical energy to bring about non-spontaneous chemical trasformations. Electrolyte.

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Plus Two Chemistry Chapter Full Notes Kerala Plus Two Chemistry Notes Chapter 3 Electrochemistry. Electrochemistry It is the study of production of electricity from I energy which is released during spontaneous chemical reactions, also the use of electrical energy to bring about non-spontaneous chemical trasformations.

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2 Fe(s) ? 2 Fe²⁺ + 4\(\overline{ | e } \)E° = -0.44 V. Electrons released at anodic spot move through metal and go to another spot on the metal and reduce O₂ in presence of H⁺. This spot behaves as cathode. O₂ (g) + 4 H⁺ + (aq) + 4\(\overline{ | e } \) ? 2 H₂ O(l) E° = 1.23 V. The overall reaction is, 2 Fe(s) + O₂ (g)+ 4H⁺ + (aq) ? 2 Fe²⁺ + 2H₂ O(l) E° = 1,67V

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Kerala Plus Two Chemistry Notes Chapter 2 Solutions. Solution A homogeneous mixture of two or more sub-stances is knows as solution. Solute : The substance present in smaller amount in a solution is called solute. Solvent: The substance present in larger amount in a solution is called solvent.

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Kerala Plus Two Chemistry Notes Chapter 9 Coordination Compounds. Coordination chemistry – branch of chemistry which deal with the complex compounds formed by tmsition and other metals. Chlorophyll, haemoglobin and vitamin B 12 are coordination compounds of Mg, Fe and Co respectively. Werner's Theory of Coordination Compounds The main postulates are,

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2Ca(OH) 2 + 2Cl 2 ? Ca(OCl) 2 + CaCl 2 + 2H 2 OCl 2 is a powerful bleaching agent. Cl 2 + H 2 O ? 2HCl + [O] Coloured substance + [O] ? colourless substance. Uses: For bleaching wood pulp, cotton and textiles; for the preparation of insecticide, pesticides and other organic solvents, e.g. CHCl 3, DDT, BHC etc. Hydrogen Chloride (HCl): Preparation:

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Nuclear chemistry comprises isotope chemistry, radiochemistry, radiation chemistry and nuclear reaction chemistry, along with applications. These interrelated fields are all covered in this textbook for chemists and chemical engineers. This new edition of the standard work 'Nuclear Chemistry' has been completely rewritten and restructured to suit teaching and learning needs in a wide range of chemistry courses, such as basic courses in radiochemistry, or more advanced nuclear chemistry courses. The book is divided into sections that closely fit teaching demands. The first chapter gives a broad introduction and background to the subject, and the second chapter covers stable isotopes. Chapters 3 to 9 comprise what is generally regarded as 'radiochemistry'. Chapters 10 to 17 offer a course in nuclear reaction chemistry. Chapter 18 deals with biological radiation effects for the chemist. The last four chapters give a guide to nuclear energy: energy production, fuel cycle, waste management, the largest applied field of nuclear chemistry. Over 200 exercises, with model answers, remain largely unchanged from the first edition, so teachers working from the earlier text should find only advantages in switching to this new restructured course book on all aspects of nuclear chemistry. The book fully meets the authors objectives, it is well written in a logical, objective, thought-provoking and quite easily readable style. It should appeal to the serious student of radio- and nuclear chemistry at either undergraduate or postgraduate level, as well as to readers with a more general interest in nuclear science and its impact on the environment.' - Applied Radiation and Isotopes, July 1995 'This book is an excellent, readable account of a significant part of the scientific achievements of more than half this century. The authors have dedicated the book to Nobel Laureate Glenn T. Seaborg and its scholarship makes it a fitting tribute.' - Radiological Protection Bulletin, December 1995

Blei and Odian's text gives students the tools they need to develop a working understanding of chemical principles—rather than just asking them to memorize facts. Now available in a new media-enhanced version, complete with its on own online course space, learning environment ChemPortal, Blei/Odian is better suited than ever to meet the needs of the students taking this course. The Media Update version of Blei/Odian includes references to dynamic, interactive tutorials, which provide a step-by-step walkthrough of concepts and problem-solving skills, as well as answer-specific feedback and practice problems. We recognize that all introductory courses are not alike. For that reason, we offer this text in three versions, so you can choose the option that's right for you: General, Organic, and Biochemistry (cloth: 0-7167-4375-2, paper: 1-4292-0994-1) – the comprehensive 26-chapter text. An Introduction to General Chemistry (0-7167-7073-3) – 10 chapters that cover the core concepts in general chemistry. Organic and Biochemistry (0-7167-7072-5) – 16 chapters that cover organic and biochemistry plus two introductory chapters that review general chemistry.

Emphasizing the applications of chemistry and minimizing complicated mathematics, GENERAL, ORGANIC, AND BIOLOGICAL CHEMISTRY, 7E is written throughout to help students succeed in the course and master the biochemistry content so important to their future careers. The Seventh Edition's clear explanations, visual support, and effective pedagogy combine to make the text ideal for allied health majors. Early chapters focus on fundamental chemical principles while later chapters build on the foundations of these principles. Mathematics is introduced at point-of-use and only as needed. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

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The Chemistry of Chlorine, Bromine, Iodine and Astatine is a special edition that contains selected sections and addresses the needs of specialists in their respective fields. The text describes the general atomic properties of non-metals, particularly the halogens, as being the perfect series to study, both in physical and chemical terms. The book explains that the combination of the atomic properties implies excellent electronegativity values for the halogen atoms. The text also cites some behavior characteristics of halogens that are irregular, such as chlorine and bromine are similar but differ from fluorine on one side and iodine on the other. The book also compares the general methods of producing chlorine, bromine, or iodine by 1) oxidation of halide derivatives or 2) reduction of compounds of the halogens in positive oxidation states. The text then reviews the application of a complex valence theory that raises difficult questions about the bonding in halogen-oxygen molecules. The book also explains the biological behavior of astatine that accumulates in the liver or in the thyroid gland depending on the method of administration either as a radiocolloid or as a true solution. The book is suitable for molecular biologists and researchers, molecular chemists, and medical researchers.

The Chemistry of Iron, Cobalt and Nickel deals with the chemistry of iron, cobalt, and nickel and covers topics ranging from the occurrence and distribution of all three elements to their properties, allotropy, and analytical chemistry. Compounds of iron, cobalt, and nickel in both low and high oxidation states are also discussed. This book is divided into three sections and begins with the history of iron, along with its occurrence and distribution, allotropy, and preparation and industrial production. The nuclear, physical, and chemical properties of iron, as well as the biological importance of iron compounds, are also considered. Compounds of iron are discussed, including carbonyls and nitric oxide complexes. The next two sections deal with the history, occurrence and distribution, allotropy, analytical chemistry, and preparation and industrial production of cobalt and nickel, along with their nuclear, physical, and chemical properties. Compounds of cobalt and nickel are examined, from carbonyls and nitrosyls to cyanides and organometallic compounds. This monograph will be a useful resource for inorganic chemists.

The Chemistry of Manganese, Technetium and Rhenium deals with the chemistry of manganese, technetium, and rhenium and covers topics ranging from the occurrence and metallurgy of all three elements to their properties and compounds. Among the compounds considered are manganese halides, cyanides, and oxides as well as carbonyls and organometallic compounds, thiocyanate complexes, and chalcogenides. This volume is divided into three sections and opens with an overview of the history and occurrence of manganese, along with its metallurgy, uses, and properties. A variety of manganese compounds are examined, including halides and cyanides, sulfides and selenides, tellurides and borates, and nitrites and nitrates. The next two sections focus on technetium and rhenium, their discovery, isolation, and general properties. Compounds of both elements are described, including hydric compounds, cyanide and thiocyanate complexes, and oxoacids and salts. Perrhenic acid and the perrhenates are also discussed, together with chalcogenides and refractory compounds, carbonyls, and organometallic derivatives. This book will be a valuable source of information for inorganic chemists.

The Chemistry of Titanium, Zirconium and Hafnium deals with the chemistry of titanium, zirconium, and hafnium and covers topics ranging from the occurrence and metallurgy of all three elements to their nuclear, physical, and chemical properties as well as analytical chemistry. The compounds of titanium, zirconium, and hafnium are also discussed. This volume is comprised of two chapters and opens with a historical overview and discovery of titanium, along with its occurrence and distribution, metallurgical aspects, and nuclear and physicochemical properties. The compounds of titanium are also considered, including alloys and complexes; hydrides and oxides; halides and oxyhalides; titanates and antimonides; and carbides and borides. The second chapter is devoted to zirconium and hafnium, their occurrence and metallurgy; and physical, chemical, and biological properties. Compounds of zirconium and hafnium are described, from alloys and hydrides to zirconates and hafnates; nitrides, phosphides, and arsenides; carbides, silicides, and germanides; molybdates, tungstates, halates, and perchlorates; alkoxides, mercaptides, and dithiocarbamates; and amides, alkylamides, triazenes, phthalocyanines, and bipyridyls. This book will be a valuable source of information for inorganic chemists.

The Chemistry of Copper, Silver and Gold deals with the chemistry of copper, silver, and gold and covers topics ranging from the occurrence and metallurgy of copper to copper compounds and compounds containing copper-metal bonds, compounds of silver, and gold alloys. Hydrides and halides, cyanides and oxides, hydroxides and oxyacids, and thiocyanates and selenocyanates are also discussed. This volume is comprised of three chapters and opens with a brief history of copper, along with its occurrence and metallurgy, analysis, and compounds. The next chapter is devoted to silver and its compounds, while the last chapter describes gold, its isotopes and alloys, chemistry, and gold hydrides and halides, cyanides and oxides, hydroxides and oxyacids. Gold sulfides, selenides and tellurides, and nitrates are also considered, along with nitrides, azides, phosphides, and arsenides; and thiosulfates, selenates, selenites, thiocyanates, and selenocyanates. The final sections look at gold complexes and the organometallic and analytical chemistry of gold. This book will be a valuable source of information for inorganic chemists.