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All the different elements are arranged in a chart called the periodic table. A Russian scientist called Dmitri Mendeleev produced one of the first practical periodic tables in the 19th

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Elements are arranged in order of
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elements are called groups; Horizontal rows of elements are called periods; Across the period, the elements change from metals to non-metals

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periodic table is a useful table that helps scientists obtain information before experimenting or testing anything. There are three main groups on the periodic table metal, nonmetal, and metalloids. Each element can be classified into one of these categories.

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The creator of the periodic table, Dmitri Mendeleev, in 1869 began collecting and sorting known properties of elements, like he was playing a game, while traveling by train. He noticed that there were

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groups of elements that exhibited similar properties, but he also noticed that there were plenty of exceptions to the emerging patterns.

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The periodic table is designed to be a

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were discovered, the periodic table was used to predict the chemical and physical properties of elements in the gaps on the table. Today, the table can be used to predict properties of elements yet to be discovered, although these new elements are all highly radioactive and break down into

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The periodic table helps predict some properties of the elements compared to each other. Atom size decreases as

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you move from left to right across the table and increases as you move down a column. The energy required to remove an electron from an atom increases as you move from left to right and decreases as you move down a column.

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The periodic table is a chart that

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Scientists use to organize the elements. The chart shows each element's name and symbol, the number of protons in its nucleus, and its characteristics. The elements are listed in order by atomic number.

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The periodic table of elements is a graphical representation of the periodic law. The periodic law dictates that elements sharing similar properties show up in groups, based upon their atomic weight (as originally proposed by Mendeleev) or, more

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accurately, atomic number.

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Elements and the periodic table.

Elements are the purest form of matter and can not be broken down into any other substance by either a physical or

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chemical change. There are about 114 elements and they are organized on a modern Periodic Table of the Elements. The Elements are organized by their atomic numbers from top to bottom and left to right.

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