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Molarity By Dilution Chemistry If8766 own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar. Molarity Chemistry If8766 Answers Pg 68 Bookmark File PDF Chemistry If8766 Molarity Answers By Dilution Chemistry If8766 Answers Page 8/25

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where the subscripts “1” and “2” refer to the solution before and after the dilution, respectively. Since the dilution process does not change the amount of solute in the solution, $n_1 = n_2$. Thus, these two equations may be set equal to one another:

$[M_1L_1 = M_2L_2]$ This relation is commonly referred to as the dilution equation.

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