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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B,
continued 17. A substance combines with oxygen, releasing a large
amount of energy as heat and light, in a(n) . 18. The decomposition of
a substance by an electric current is called . 19. A(n) orders the
elements by the ease with which they undergo certain chemical
reactions. 20.

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B,
continued 18. When a strong acid is titrated with a weak base, the pH

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of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry 137 Chapter Test Name Class Date Chapter Test A, continued ____ 19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. ____ 20. In the figure on the previous page, the volume of titration standard

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. ____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in
...

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18. Dalton's atomic theory agreed with the modern atomic theory EXCEPT for the statement that all atoms of the same element have the same Complete the following table to compare the types of subatomic particles. 19. 20. 21. 10 Particle Proton Neutron Electron CHAPTER 3 TEST Mass number Relative charge Location MODERN CHEMISTRY

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STUDY. PLAY. The basis of an ionic bond is the__ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a__ noble gas. Ions that are formed from more than one atom are called__

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