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NOTES S - block

elements part - 2

(chemical properties)

with complete notes.

Characteristics of d

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elements ll #Class11 ll

CHEMISTRY II Piano

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And Their Properties
Note

Properties of Nonmetals

- Phosphorus, sulfur, and chlorine are among these other elements found in your body.
- These elements are classified as nonmetals.

19.2 Nonmetals

- Nonmetals are

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elements that usually are gases or brittle solids at room temperature.

Chapter 19: Elements and Their Properties

Each element has its own unique properties. Each contains a different number of protons and neutrons, giving it its own atomic number and mass number. The atomic

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Number of an element is
equal to the number of
protons that element
contains. The mass
number is the number of
protons plus the number
of neutrons of that
element. Therefore, it is
possible to determine
the number of neutrons
by subtracting the
atomic number from the
mass number.

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The metallic character is used to define the chemical properties that metallic elements present. Generally, metals tend to lose electrons to form cations. Nonmetals tend to gain electrons to form anions. They also have a high oxidation potential

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therefore they are easily oxidized and are strong reducing agents.

Periodic Properties of the Elements -

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Properties of Atoms and
the Periodic Table

Structure of the Atom

H, He, B, Be, I, F, Br,

Es Lr. Physical Science.

Notes. Properties of

Get Free Elements And Their Properties Atoms and the Periodic Table. Structure of the Atom. Elements are abbreviated in scientific shorthand. • Symbols

on the periodic table are short or abbreviated ways to write the name of an element.

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Family.Group 17 - The
Halogens ...

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...**

Properties of elements
are the periodic function
to their atomic numbers.
The periodicity in

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properties is due to
repetition of similar
outer shell electronic
configuration at a
certain regular intervals.

In modern periodic table
is based on modern
periodic law in which
elements are arranged in
increasing order of their
atomic numbers.

Classification of Elements &

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Modern periodic law:

This law is based on the assumption that the physical and chemical properties of elements are the periodic function of their atomic number.

Characteristics: The 9 vertical columns present were called group. 1st to VIIIth group + 0 group of inert gases.

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Ramsay introduced the
inert gases. The 7
horizontal series termed
as periods.

Answers

**Periodic Classification
of Elements: Notes for
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elements. substances
with only one type of
atom. metal. an element
that is malleable,
polishes to a luster,
conducts heat and

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electricity well, and
tends to lose electrons in
a chemical reaction.
metalloid. an element
with properties of both
metals and nonmetals.
nonmetal.

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their properties
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Chemistry Chapter 3 •
Genesis of Periodic
Classification

Dobereiner's Triads In
1829, Dobereiner
arranged certain
elements with similar
properties in groups of
three in such a way that
the atomic mass of the
middle element was
nearly the same as the
average atomic masses

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of the first and the third
elements.
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Element families are
indicated by numbers

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located at the top of the periodic table. Todd Helmenstine. An element family is a set of elements sharing common properties.

Elements are classified into families because the three main categories of elements (metals, nonmetals, and semimetals) are very broad. The characteristics of the

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elements in these
families are determined
primarily by the number
of electrons in the outer
energy shell.

**Element Families of
the Periodic Table -
ThoughtCo**

Mendeleev predicted the
existence and properties
of Germanium (Ge) and
Gallium (Ga). When
they were discovered,

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their properties were
found to closely match
those predicted by
Mendeleev; 1913:

Henry Moseley
discovers concept of
atomic numbers.
Observed that
frequencies of Xrays
were different for each
element.

7.S: Periodic
Properties of the

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arrange the elements
originally by atomic
weight in a scheme that
would show similarity
among groups"what are
the answers for chapter
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2018 - what are the
answers for chapter 19
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on the elements and
their properties' 'import
spring config file based
on property in properties

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Presently, there are 118
elements known to us.

Out of these 118

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chemical elements,
some elements are man-
made. When there were
only 31 elements it was
relatively easy to study
the properties of these
chemical elements
individually.

**Periodic Classification
of Elements - History,
Periodic ...**

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Answers

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Meyer presented the
classification of
elements in the form of
a curve between atomic
volume and atomic
masses and state that the

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properties of the elements are the periodic functions of their atomic volumes.
[Here, atomic volume = molecular mass / density
He concluded that the elements with similar properties occupy similar position in the curve.

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