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AP Chemistry Chapter 4. Aqueous

Reactions and Solution Stoichiometry - 2 -

Ionic Compounds in Water □ When an

ionic compound dissolves in water, the

ions are said to dissociate. □ This means

that in solution, the solid no longer exists

as a well-ordered arrangement of ions in

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Compounds in Water □ When an ionic
compound dissolves in water ...

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Read Online Ap Chemistry Aqueous Solutions Canicu Ap Chemistry Aqueous Solutions Canicu 4.1 General Properties of Aqueous Solutions □ A solution is a homogeneous mixture of two or more substances. □ A solution is made when one substance (the solute) is dissolved in another (the solvent). □ The solute is the substance that is present in ...

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CHEMISTRY FREE-RESPONSE

QUESTIONS . GO ON TO THE NEXT

PAGE. -6-(c) Calculate the concentration of urea, in mol/L, in the saturated solution at 20.°C. (d) The student also determines that the concentration of urea in a saturated solution at 25°C is 19.8 . M.

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2. Placing aluminum foil in any solution containing aqueous copper(II) ions will result in a reaction. The reaction is slow to begin with, then proceeds rapidly. a) Referring to a solubility table, name at least four ionic compounds that could be dissolved in water to make a solution containing aqueous copper(II) ions.

AP Chemistry - Molecular, Ionic and Net Ionic Equations

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